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## Key indicators

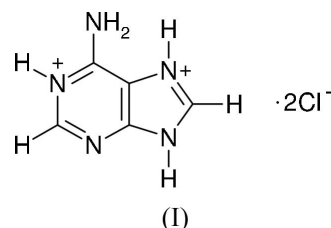
Single-crystal X-ray study  
 $T = 150$  K  
Mean  $\sigma(C-C) = 0.002$  Å  
 $R$  factor = 0.024  
 $wR$  factor = 0.063  
Data-to-parameter ratio = 11.4For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.Redetermination of adeninium dichloride:  
the question of centrosymmetry

The low-temperature redetermination of adeninium(2+) dichloride,  $C_5H_7N_5^{2+} \cdot 2Cl^-$ , obtained as part of an experimental polymorph screen on adenine, is reported here. The crystal structure is shown to be centrosymmetric. Cations and anions are connected through  $N-H \cdots N$  and  $N-H \cdots Cl$  hydrogen bonds [ $N \cdots N = 2.899(2)$  Å and  $N \cdots Cl = 3.0274(14)$ – $3.5155(16)$  Å] to form sheets perpendicular to the  $b$  axis.

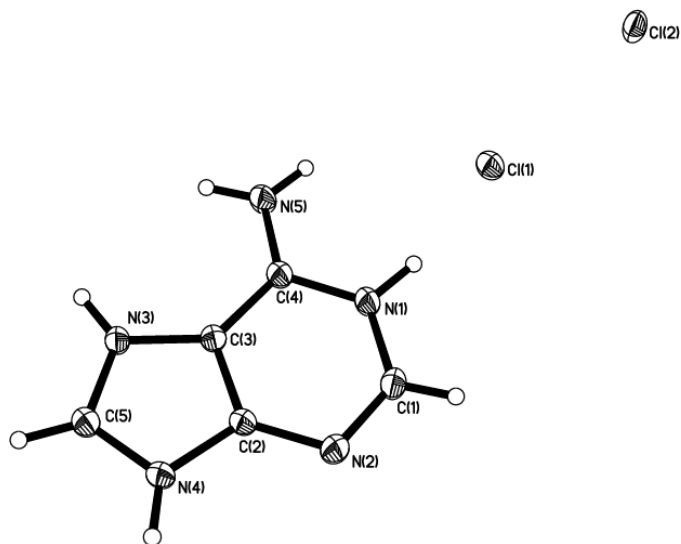
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## Comment

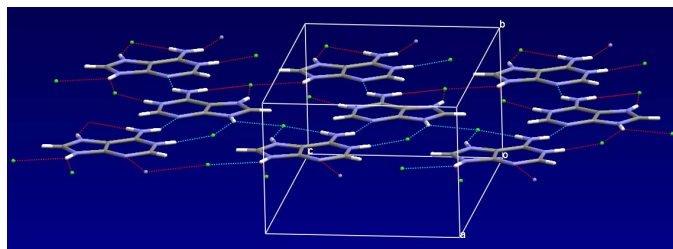
The title compound, (I), is a hydrochloride salt of adenine, which is one of the two common purine bases found in ribose and deoxyribose nucleic acids.



The unit cell was determined in 1974 (Iwasaki, 1974); however, it was not possible unequivocally to establish the correct space group, either  $Pna2_1$  or  $Pnam$  (non-standard setting of  $Pnma$ ), as refinement in each gave similar  $R$  values (0.043 and 0.045, respectively). The structure was also determined at room temperature by Kistenmacher & Shigematsu (1974), and refined in the centrosymmetric space group  $Pnma$ , giving an  $R$  value of 0.035. In this space group, mirror symmetry is imposed on the adenine dication, with some atoms having large r.m.s. displacements normal to the mirror plane. However, it was argued that purines commonly show some bending about the  $C2-C3$  bond axis (Sletten & Jessen, 1969), which is inconsistent with the analysis in the centrosymmetric space group. Hence, it was suggested that the true space group could be  $Pn2_1a$  (non-standard setting of  $Pna2_1$ ). We have redetermined the crystal structure at 150 K, to gain more precise data for our molecular modelling studies. The structure was refined in both  $Pnma$  and  $Pna2_1$ , giving  $R$  values of 0.0241 and 0.0229, respectively, despite the statistical averages for the normalized structure factors ( $E$  values) being more consistent with a centrosymmetric than a non-centrosymmetric distribution. However, when refined in the non-centrosymmetric space group, all the ring H atoms deviate by between  $13-15^\circ$  from the mean ring plane to which they are attached. These are large deviations when compared with other adeninium crystal structures, which include adeninium



**Figure 1**  
View of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 50% probability level.



**Figure 2**  
View of the hydrogen-bonded sheet motif present in (I), with the hydrogen bonds shown as dotted lines;  $D\cdots A$  distances greater than 3.3 Å have been omitted for clarity.

sulfate (Langer & Huml, 1978), adeninium dinitrate (Hardgrove *et al.*, 1983) and adeninium diperchlorate monohydrate (Bendjedou *et al.*, 2003). In addition, analysing the non-centrosymmetric structure with *PLATON* (Spek, 2003) to search for missing or higher symmetry gave the centrosymmetric structure at 100% confidence level. Hence, using the superior low-temperature data, we can conclude that the most likely space group of (I) is *Pnma*.

In this low-temperature determination, the precision of the unit-cell dimensions was improved by an order of magnitude, and the unit-cell volume decreased by *ca* 14 Å<sup>3</sup>, consistent with the determination at low temperature. In general, the metric parameters are not significantly different, within standard deviations, from those found at room temperature. The adenine molecule is protonated at N1 and N3, with the C–N bond lengths in the rings in the range 1.308 (2)–1.375 (2) Å, and the C2–C3, C3–C4 and C4–N5 bond lengths being 1.379 (2), 1.409 (2) and 1.310 (2) Å, respectively. In the crystal structure, the cations are linked through N–H $\cdots$ N hydrogen bonds to form extended chains in the *a*-axis direction. These

chains are, in turn, linked by N–H $\cdots$ Cl hydrogen bonds to form sheets (Fig. 2) lying parallel to the (040) family of lattice planes. Four of the H atoms on the adenine cation are involved in N–H $\cdots$ Cl hydrogen bonds (see Table 1) and, in addition, atoms H4 and H6 are involved in weaker bifurcated N–H $\cdots$ Cl hydrogen bonds, with N $\cdots$ Cl distances of 3.2936 (15) and 3.5155 (16) Å, respectively. There are two independent Cl<sup>−</sup> ions within the hydrogen-bonded sheets: Cl1, which is involved in one conventional and three weaker bifurcated N–H $\cdots$ Cl hydrogen bonds, and Cl2, which is involved in three conventional N–H $\cdots$ Cl hydrogen bonds. In the N–H $\cdots$ N and N–H $\cdots$ Cl hydrogen-bonded sheets, all acceptors and donors are used.

## Experimental

As part of an experimental polymorph screen on adenine, (I) was obtained by evaporation of a solution of equimolecular amounts of thymine/adenine, and cytosine/adenine in dilute hydrochloric acid, giving colourless block-shaped crystals.

### Crystal data

C<sub>5</sub>H<sub>7</sub>N<sub>5</sub><sup>2+</sup>·2Cl<sup>−</sup>  
*M<sub>r</sub>* = 208.06  
 Orthorhombic, *Pnma*  
*a* = 13.4405 (11) Å  
*b* = 6.4774 (5) Å  
*c* = 9.3684 (7) Å  
*V* = 815.61 (11) Å<sup>3</sup>  
*Z* = 4  
*D<sub>x</sub>* = 1.694 Mg m<sup>−3</sup>

Mo *K*α radiation  
 Cell parameters from 5209 reflections  
 $\theta$  = 2.7–28.1°  
 $\mu$  = 0.74 mm<sup>−1</sup>  
*T* = 150 (2) K  
 Block, colourless  
 0.74 × 0.26 × 0.24 mm

### Data collection

Bruker SMART APEX diffractometer  
 Narrow-frame  $\omega$  scans  
 Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)  
 $T_{\min}$  = 0.609,  $T_{\max}$  = 0.842  
 6736 measured reflections

1076 independent reflections  
 1064 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}}$  = 0.016  
 $\theta_{\max}$  = 28.3°  
 $h$  = −17 → 17  
 $k$  = −8 → 8  
 $l$  = −12 → 12

### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)]$  = 0.024  
 $wR(F^2)$  = 0.063  
 $S$  = 0.99  
 1076 reflections  
 94 parameters  
 All H-atom parameters refined

$w = 1/[\sigma^2(F_o^2) + (0.0337P)^2 + 0.5379P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} < 0.001$   
 $\Delta\rho_{\max} = 0.37 \text{ e } \text{Å}^{-3}$   
 $\Delta\rho_{\min} = -0.25 \text{ e } \text{Å}^{-3}$

**Table 1**

Hydrogen-bonding geometry (Å, °).

<i>D</i> –H $\cdots$ <i>A</i>	<i>D</i> –H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> –H $\cdots$ <i>A</i>
N1–H2 $\cdots$ Cl1	0.94 (3)	2.11 (3)	3.0274 (14)	167 (2)
N3–H3 $\cdots$ Cl2 <sup>i</sup>	0.89 (3)	2.25 (3)	3.0693 (14)	153 (2)
N4–H4 $\cdots$ Cl1 <sup>ii</sup>	0.90 (2)	2.53 (2)	3.2936 (15)	143.8 (19)
N4–H4 $\cdots$ Cl2 <sup>ii</sup>	0.90 (2)	2.56 (2)	3.1695 (14)	126.1 (18)
N5–H6 $\cdots$ N2 <sup>iii</sup>	0.85 (2)	2.28 (2)	2.899 (2)	129.6 (19)
N5–H6 $\cdots$ Cl1	0.85 (2)	2.82 (2)	3.5155 (16)	140.3 (18)
N5–H7 $\cdots$ Cl2 <sup>i</sup>	0.88 (3)	2.22 (3)	3.0985 (16)	175 (2)

Symmetry codes: (i)  $x, y, 1 + z$ ; (ii)  $\frac{1}{2} + x, y, \frac{3}{2} - z$ ; (iii)  $x - \frac{1}{2}, y, \frac{3}{2} - z$ .

H atoms were refined independently using an isotropic model.

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINTE* (Bruker, 2000); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 2000) and *MERCURY* (Bruno *et al.*, 2002); software used to prepare material for publication: *SHELXL97*.

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