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## **4-Amino-5-fluoropyrimidin-2(1H)-one–2-amino-5-fluoropyrimidin-4(3H)-one–water (1/1/1)**

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## Key indicators

Single-crystal X-ray study

T = 150 K

Mean  $\sigma(\text{C}-\text{C}) = 0.002 \text{ \AA}$ 

R factor = 0.044

wR factor = 0.123

Data-to-parameter ratio = 11.3

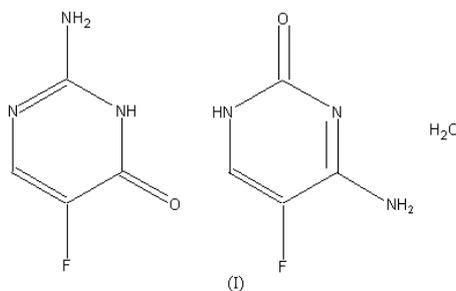
For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.4-Amino-5-fluoropyrimidin-2(1H)-one–  
2-amino-5-fluoropyrimidin-4(3H)-one–  
water (1/1/1)

The title co-crystal,  $\text{C}_4\text{H}_4\text{FN}_3\text{O} \cdot \text{C}_4\text{H}_4\text{FN}_3\text{O} \cdot \text{H}_2\text{O}$ , has one molecule of 4-amino-5-fluoropyrimidin-2(1H)-one, one molecule of its isomer 2-amino-5-fluoropyrimidin-4(3H)-one and a molecule of water in the asymmetric unit. 4-Amino-5-fluoropyrimidin-2(1H)-one is commonly known as 5-fluorocytosine.

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## Comment

The title co-crystal, (I) (Fig. 1), was grown by evaporation of a 50% aqueous solution of ethanol saturated with 5-fluorocytosine. Two different crystal forms were obtained from this solution. The major crystallisation product exhibited a block morphology and was the known monohydrate of 5-fluorocytosine (Louis *et al.*, 1982). A small number of needle-shaped crystals were observed as the minor crystallization product. These crystals proved to be the co-crystal, (I). The isomer of 5-fluorocytosine was assumed to have been present in the commercial sample of 5-fluorocytosine purchased from Fluorochem (98% pure, Old Glossop, UK) that was used to prepare the initial solution.



The simplest hydrogen-bonded subunit observed is a two-molecule unit, containing one molecule of each isomer. Each molecule of 5-fluorocytosine forms three hydrogen bonds to a

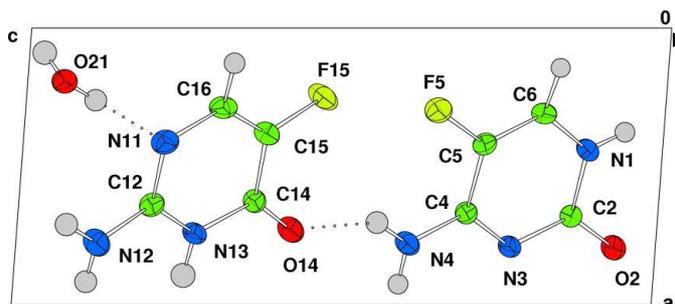
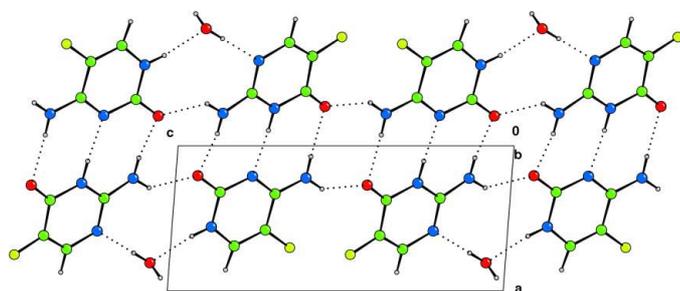


Figure 1

The asymmetric unit of the title co-crystal. Displacement ellipsoids are drawn at the 50% probability level. H atoms are shown as spheres. Dotted lines indicate hydrogen bonds.



**Figure 2**

The hydrogen bonded ribbon present in the title structure. Dotted lines indicate hydrogen bonds.

molecule of the isomer (N4—H2···O14, N13—H13···N3 and N12—H12···O2), forming two adjoining  $R_2^2(8)$  hydrogen bond rings (Table 1). Two different  $R_2^4(8)$  hydrogen-bond rings join these subunits together to form a ribbon (Fig. 2).

The role of the water molecules in the structure is to join together the ribbons into a hydrogen-bonded sheet. The water hydrogen bonds to two molecules from one ribbon, acting both as donor and acceptor, and as a donor to a third molecule, from a different ribbon (Table 1). The ribbons form stepped sheets, parallel to the  $01\bar{1}$  planes (Fig. 3).

Within the ribbon structure, there is also a close F···F contact, between F5 and F15, of 2.9003 (15) Å; however, this is likely to have arisen as a consequence of the adjacent  $R_2^4(8)$  hydrogen-bond ring.

## Experimental

Crystals were grown from a 50% aqueous ethanol solution, by evaporation at room temperature. The crystal form reported was the minor crystallisation product.

### Crystal data

$C_4H_4FN_3O \cdot C_4H_4FN_3O \cdot H_2O$   
 $M_r = 276.22$   
 Triclinic,  $P\bar{1}$   
 $a = 5.4122$  (16) Å  
 $b = 8.447$  (2) Å  
 $c = 12.083$  (4) Å  
 $\alpha = 89.454$  (5)°  
 $\beta = 85.718$  (5)°  
 $\gamma = 77.096$  (4)°  
 $V = 536.9$  (3) Å<sup>3</sup>

$Z = 2$   
 $D_x = 1.708$  Mg m<sup>-3</sup>  
 Mo  $K\alpha$  radiation  
 Cell parameters from 1511 reflections  
 $\theta = 3.0$ – $28.1$ °  
 $\mu = 0.16$  mm<sup>-1</sup>  
 $T = 150$  (2) K  
 Needle, colourless  
 $0.44 \times 0.14 \times 0.11$  mm

### Data collection

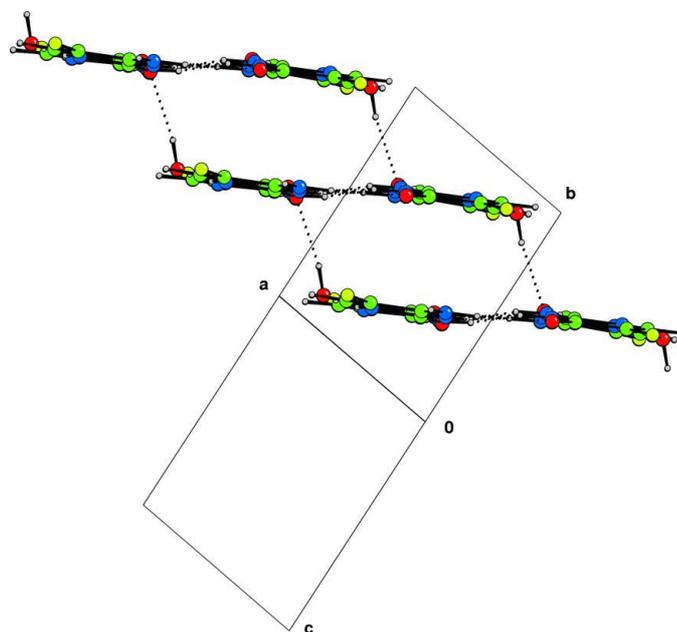
Bruker SMART APEX diffractometer  
 $\omega$  scans  
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)  
 $T_{\min} = 0.934$ ,  $T_{\max} = 0.984$   
 4532 measured reflections

2405 independent reflections  
 1884 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.018$   
 $\theta_{\text{max}} = 28.3$ °  
 $h = -6 \rightarrow 6$   
 $k = -11 \rightarrow 10$   
 $l = -15 \rightarrow 15$

### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.044$   
 $wR(F^2) = 0.123$   
 $S = 1.05$   
 2405 reflections  
 212 parameters  
 All H-atom parameters refined

$w = 1/[\sigma^2(F_o^2) + (0.0762P)^2 + 0.0364P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.36$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.24$  e Å<sup>-3</sup>



**Figure 3**

The stepped structure of the sheet, comprising ribbons which are hydrogen bonded (dotted lines) via water molecules.

**Table 1**

Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
N12—H11···O2 <sup>i</sup>	0.86 (3)	2.13 (3)	2.870 (2)	143 (2)
N12—H12···O2 <sup>ii</sup>	0.91 (2)	1.99 (2)	2.889 (2)	172 (2)
N13—H13···N3 <sup>iii</sup>	0.91 (3)	2.01 (3)	2.922 (2)	175 (2)
N1—H1···O21 <sup>iii</sup>	0.83 (2)	1.95 (2)	2.775 (2)	173 (2)
N4—H2···O14 <sup>ii</sup>	0.88 (2)	2.07 (2)	2.9482 (19)	177 (2)
N4—H3···O14	0.91 (3)	2.01 (2)	2.8285 (19)	149 (2)
O21—H21···N11	0.84 (3)	1.94 (3)	2.785 (2)	177 (2)
O21—H22···O2 <sup>iv</sup>	0.81 (3)	2.03 (3)	2.826 (2)	170 (2)

Symmetry codes: (i)  $x, y + 1, z + 1$ ; (ii)  $-x + 2, -y, -z + 1$ ; (iii)  $x, y - 1, z - 1$ ; (iv)  $x - 1, y + 1, z + 1$ .

All H atoms were located [ $C-H = 0.95$  (2)– $0.96$  (2),  $N-H = 0.83$  (2)– $0.91$  (3) and  $O-H = 0.95$  (2)] in a difference map and were refined isotropically.

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1998); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: CAMERON (Watkin *et al.*, 1996); software used to prepare material for publication: SHELXL97.

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